

PROFORMA

DETAILS OF ENTRANCE TEST – 2015-2016

Name of the Faculty: Architecture & Ekistics

Department/Centre: Architecture

Name of the Program: Bachelor of Architecture

About Program's Prospects:

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Summary of Entrance Test

S.No.	Test-Component	Test Duration (in minutes)	Max. Marks	Passing Marks	Negative Marking (Yes/No)
	Part-A PCM- Objective type	2 Hours	100	40	No
	Part-B Aptitude Test-Subjective(Illustrative) as well as objective type	3 Hours	100	40	No

Any other information about the Entrance Test:

Important Instructions for Test (Pl. add/modify as required)

Permissible Material/equipment for Entrance Test (as required):

- Black/Blue Ball Pen,
- Calculator,
- Pencil

Detailed Syllabus for the Entrance Test: Attached

Syllabus for B.Arch Entrance Test

SYLLABUS FOR P.C.M.

Physics:

Units and dimensions, SI units, displacement, velocity, kinematics in one and two dimensions with constant acceleration, projectiles, concepts of relative motion, circular motion. Newton's Laws of motion, concepts of inertial frame, momentum and energy, universal law of gravitation, variation of the acceleration due to gravity with altitude and latitude. Kepler's Law of motion of planets and satellites assuming circular orbits, simple harmonic motion. Centre of mass of a system of particles, elastic and inelastic collision in one dimension, rigid bodies, moments of inertia of simple shapes like ring, disc, cylinder and sphere, angular momentum, torque, conservation of angular momentum. Hook's law, Young's Modulus shear and bulk moduli, principle of buoyancy, pressure in a fluid wave motion, concepts of amplitude, frequency and phase, longitudinal and transverse waves, progressive and stationary waves, vibration of string and air columns, resonance, Doppler's effect, Thermal expansion of solids, liquids, and gas, gas laws, absolute temperature, specific heat, C_p , C_v and isothermal and adiabatic processes, calorimetry, latent heat, equivalence of heat and work, conduction, convection and radiation, Elements of Kinetic Theory of gases, pressure of an ideal gas equipartition of energy, coulomb's law, electric field, lines of force, electric potential. capacitors in series and parallel, energy stored in a capacitor, electric current, Ohm's law, Series and parallel arrangement of resistances and cells, Krichoff's Law, Wheatstone bridge and application, heating effect of current, current carrying wire in a magnetic field, moving coil, galvanometer and ammeter, Amphere's law, Faraday's Law, Lorentz Force, effect of magnetic field on current carrying conductors, Maxwell's equations, curved mirrors and thin lenses, magnification, microscope telescope, wave nature of light, interference phenomenon, Young's double slit experiment, fringe width, polarisation of light, cathode rays radioactivity law of decay, half life, photo-electric effect, Byohr's theory of hydrogen like atoms, X-rays production and properties, diode rectification and triode amplification, Atomic nucleus, binding energy, nuclear energy by fission and fusion, element4ary properties of semi conductors and devices based on them, Michelson Morley experiment, postulates of special theory of relativity.

Chemistry:

General Chemistry/ Physical Chemistry: Atomic mass, extraction of metals, gas laws, ideal gas equation, Kinetic molecular theory of gases, constituents of the atom, nuclear model of atom, Bohr's atomic model (Mathematical details excluded), Quantum mechanical model of atom. Aufbau principle, Hund's Rule. Paul's exclusion principle, de Broglie equation, Uncertainty principle, shapes of orbitals, Modern periodic table, periodic trends in properties, ionization energy, electron affinity, atomic radii and electronegativity, Chemical bonds, quantum theory of covalent bonds, hydrogen bond, hybridization involving s.p. & d orbital, shapes of molecules, VSEPR Concept, molecular orbital theory, solid state, X-ray studies of crystals, packing in crystals, crystal lattices, Internal energy, enthalpy, Hess's Law, entropy, free energy, heat of reactions, concept of chemical equilibrium, Law of Equilibrium, Le Chatelier's Principle and its applications, ionisation of electrolytes, acid base equilibrium, solubility product, ionization of water, pH, redox reactions, Galvanic cell and its EMF, electrode potential, effects of temperature and concentration, electrolysis, Oxidation number, Solutions, lowering of vapour pressure, Raoult's Law, chemical kinetics, rates of reaction, its dependence on concentration, order of a reaction, factors affecting rate of a reaction, rate constant, half life period of first order reaction, absorption, colloidal state, preparation and properties of colloidal solutions, gold no, radioactivity, half life, Carbon dating, chemical thermodynamics.

Inorganic Chemistry: Preparation and Properties of the following : Hydrogen, Oxygen, Halogens, Noble gases, Water, Hydrogen peroxide, Nitrogen, Boron, Silicon, Phosphorous, Sulphur, Sodium, Potassium,

Magnesium, Calcium, Aluminium, Iron, Copper, Silver, Gold, Zinc, Mercury, Tin, Lead, Transition elements, Inner transition elements, Cement, Glass, Plastics, Soaps, Detergents, Coordinate compound.

Organic Chemistry: Organic radicals, ions, inductive, electrometric and resonance effects, isomerism, Homologous series. General preparation and functional group properties of the following with emphasis on points noted against each of them.

- (a) Alkanes (free radical substitution) alkenes (electrophilic addition, Markownikoffs addition, peroxide effect) and alkynes
- (b) Alcohol, distinctions, haloform reaction
- (c) Alkyl halides (Nucleophilic substitution)
- (d) Carbonyl compounds (Nucleophilic, Addition, polymerization and condensation)
- (e) Ethers and carboxylic acids
- (f) Amines (classification, distinction, basic nature)
- (g) Benzene (resonance), Toluene, Phenol, glycerol, Oils and Fats; Proteins, nucleic acids. Carbohydrates and lipids.

Mathematics:

Set Theory and Higher Algebra -Algebra of sets, relation, One, Many one, Onto and Into Mappings, Inverse mapping, Composite mapping, Bounded and unbounded sets Neighbourhood of a point, domain, Codomain, Range. Square roots of polynomial quadratic surds, unity, A.P., H.P., G.P. and other related miscellaneous series, A.G.P., Method of differences, Sum of the squares and cubes of sum of the squares and cubes of first n nature numbers, Simple geometrical problems of permutations and combinations, General terms in the expansions of Logarithmic, Exponential series, binomial and multinomial theorems, remainder theorem, Factor theorem, rationalising factor of binomial surd of different orders, Conversion of a fraction and a quadratic surds into a continued fraction, Inequalities. Periodic function, hyperbolic function, Inverse hyperbolic functions, Greatest integer functions, Absolute function, Signum function, Even and Odd functions, Transcendental functions, Rational and Irrational algebraic functions etc.

Vector Algebra -Linear combination of vectors, linear dependence and independence of vectors, Geometrical applications of dot (scalar) and cross (vector) product of two, three and four vectors.

Plane and Solid Geometry - Classification of curves represented by general polynomial equations (homogeneous and non-homogeneous) of second degree in x and y , Homogeneous equations of n th degree, Detailed study of families of Straight lines and Circles, Parabola, Ellipse and Hyperbola, condition of tangency of a line to a curve of any positive integral degree in x and, simple problems on plane, straight line in space and sphere.

Trigonometry - Trigonometrical conditional identities and equations, Relations between sides, angles and radii of incircle, circum circle and excribed circle associated with triangle, Inverse circular functions, Polar form of complex number, Euler's formula, Values of $(a + ib)^{p/q}$, Applications of De-Moivre's theorem.

Determinants and matrices - Multiplications of determinants and matrices, adjoint and inverse of matrices, rank of matrix, Cramer's rule, Applications of determinants and matrices, Geometrical transformations (i.e. reflection, rotation, translation, enlargement), their composition and representation by matrices. Limit, Continuity, Monotonocity, Differentiability, Differentiation of composite function, Implicit relation, Parametric forms, logarithmic differentiation, differentiation by transformations, increasing and decreasing functions, Tangent, Normal, Maxima and Minima of a function of one variable, velocity, Acceleration, Rolle's theorem, Lagrange's mean value theorem, standard results on successive (ordinary and partial) differentiation. L'Hospital's rule, Seven indeterminate forms, Euler's theorem on homogenous function for partial differentiation, Simple problems on Leibnitz's theorem. Equations of tangent plane and normal line at any point of 3 D figures.

Integral Calculus - Integration by well known standard results, by substitutions, by parts, by partial fractions, some well known definite integrals, properties of definite integrals, Reduction formulae, Applications of definite

integral in finding the area of bounded region in x - y plane, Volume and lateral (curved) surface area of some well known 3-D solids obtained by revolution, Mean value, Mean square value, root mean square value of a function for a given interval, Calculation of a interval in which the value of a given definite integral exists.

Differential Equations - Formation of ordinary and partial differential equations, General and particular solution of O.D.E. of I order and I degree (i.e. variable separable form, homogeneous form, linear form, exact form and their reducible form obtained by some transformations and integrating factors), O.D.E. of I order but not of I degree, General solution of $\frac{dy}{dx} = G(x)$, $\frac{d^2y}{dx^2} = H(y)$. Karl Pearson's simple correlation coefficient for a tabulated function between two variable Coefficients of covariance and variation, Lines of Regression, regression coefficients Properties of binomial, normal and poisson distributions. Measures of skewness and Kurtosis.

SYLLABUS FOR APTITUDE TEST

Reasoning and Awareness (Objective Type)

Current affairs- Science & Technology, Sports & Entertainment, Place & Buildings, Personalities etc.

Social and Cultural Awareness.

Reasoning- Simple reasoning to select the odd one out from the set of entities, to find sequence in progression.

Drawing (Illustrative Type)

Free Hand Sketching- Free hand sketching of simple objects from everyday life such as furniture, kitchenware, vehicles and natural elements such as flowers, leaves, plants, trees etc from observation and recall from memory. Sense of Scale and proportion.

Enlargement & Reduction- Enlargement and reduction of given figures, understanding of scale and proportion.

Geometrical Drawing- Drawing of basic geometrical shapes such as lines, angles, triangles, polygons, circles etc and their compositions.

Abstract Drawing- Exercises to complete an abstract figure provided into a creative pattern or art based on visualisation, imagination and creativity.

Sketching of Scene- Sketching of scenes from everyday urban and rural life commonly observed such as markets, festival, mela, ghats, playgrounds, railway stations, bus stops etc; from memory. Ability to draw objects from imagination with details like colour, texture, materials etc;

Colouring- Colouring in pencil, water/poster colours or any other medium for colouring techniques and composition of colours.